Supplying Nutrients to Crops

- OPlants need nutrients for healthy growth and development.
- OPlant nutrition involves the absorption of nutrients for plant growth and is dependent on 16 essential elements, often referred to as nutrients.



- OThree of the 16 elements comprise 89 percent of a plant's tissue by dry weight: oxygen (O), hydrogen (H), and carbon (C).
- OThey are considered to be non-fertilizer nutrients as we can NOT buy these.
- OThe plant acquires these three elements through natural processes from air and water.

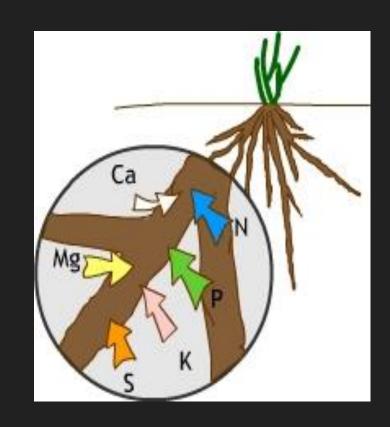
Six essential elements in addition to oxygen, hydrogen, and carbon are required in greater quantity than the others; they are called macronutrients.

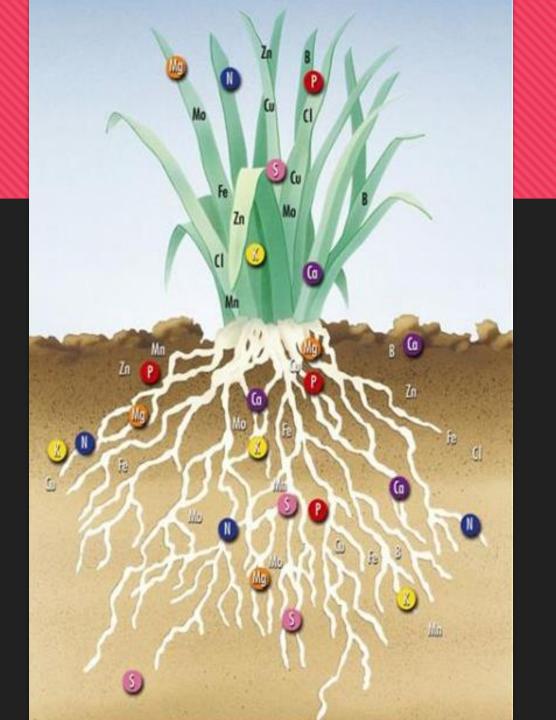
OThe macronutrients are nitrogen (N), phosphorus (P), potassium (K), calcium (Ca), magnesium (Mg), and sulfur (S).

- 1. Nitrogen, phosphorus, and potassium are considered primary macronutrients because they are used in complete fertilizers.
 - ONitrogen promotes green leafy growth.
 - OPhosphorus encourages flowering and root growth.
 - OPotassium provides disease resistance.

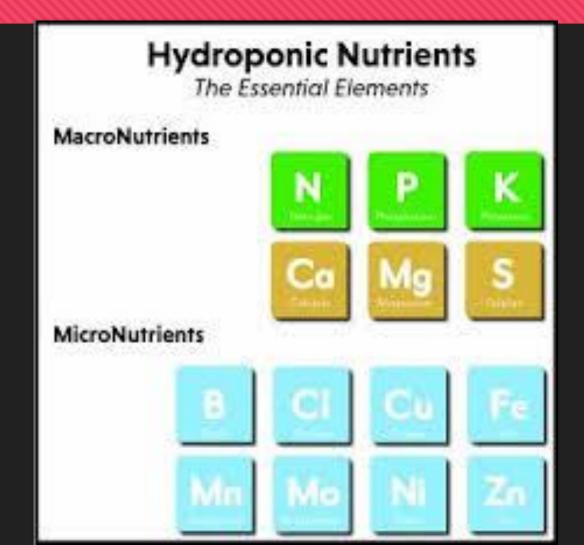
O2. Calcium, magnesium, and sulfur are said to be secondary macronutrients because plants need them in moderate amounts.

OThese secondary macronutrients may or may not be used in complete fertilizers.





- OThe other seven essential elements, called micronutrients, are needed in small quantities.
- OThese are sometimes called trace elements. They are boron (B), copper (Cu), chlorine (Cl), iron (Fe), manganese (Mn), molybdenum (Mo), sodium (Na), and zinc (Zn).
- OSix of these (boron, copper, iron, manganese, molybdenum, sodium, and zinc) are supplied to plants as fertilizers.
- OChlorine is not added to fertilizers since plants obtain sufficient quantities of chlorine from the medium or from water.



A little phrase can be used to help memorize the 16 essential elements for plant growth.

- Olt is "C. B. Hops Café Mighty Good, Closed Monday, See You Zen."
- Olt represents the following: Carbon (C), Boron (B), Hydrogen (Hops), Oxygen (HOpkins), Phosphorus (HoPkins), Potassium (HopKins), Nitrogen (HopkiNs), Sulfur (HopkinS), Calcium (Café), Iron (café), Magnesium (Mighty good), Chlorine (Closed), Manganese (Monday), Molybdenum (Morning), Copper (See you = Cu), Zinc (Zen).

- Other elements play important roles in plant growth and development.
- OFor instance, silicon (Si) improves plant

strength and disease resistance.

ONickel (Ni) is another element considered important for plant growth.

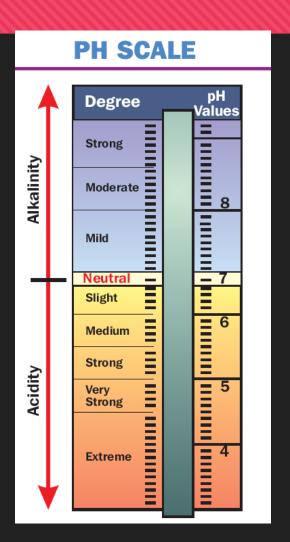
- OPlants receive most of the nutrients they need from the growing media.
- OIn order to maintain healthy plants, a grower must provide the right type and amount of nutrients to the media so the plants can absorb the nutrients and grow.
 - O1. A water test can be performed to determine which nutrients are present and which nutrients are deficient or lacking. (Armstrong City Water)
 - O2. When nutrients are deficient in the water, the plant growth is adversely affected.

- Olt is common for plant leaves to show symptoms of a *nutrient deficiency* by turning colors.
- O When nitrogen is deficient in the soil, a plant's older leaves turn yellow.
- O Abnormal yellowing of plant leaves is a condition called *chlorosis*.
- O A plant will show purpling in the stem or leaf when phosphorus is deficient.



- OMany times, the nutrients needed for plant growth are present in our water reservoir, but the plants do not have access to the nutrients.
- ONutrient availability is influenced by the pH in your nutrient water.
- pH measures the amount of acidity or alkalinity and is based on the amount of hydrogen ions present in the water.

- OSoil pH can range from 1 to 14.
- OA pH reading of 7 is neutral.
- OSubstances that have pH readings below 7 are acidic.
- OSubstances with pH readings above 7 are alkaline or basic.



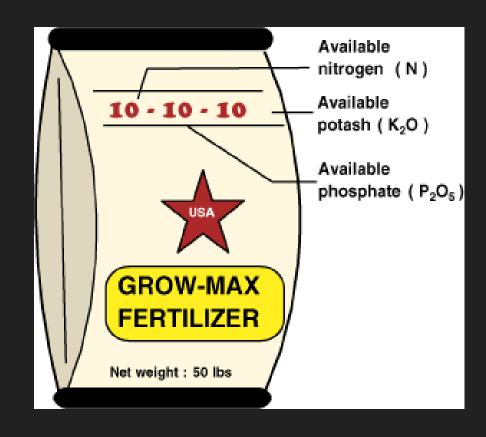
- OPlants have specific pH ranges that are ideal for maximum plant growth.
- OMost plants grow best at a pH of 5.8 to 6.4 because the most nutrients are available for the plant to absorb in that range.
- OSome important horticultural plants (azaleas) do better in a medium that has a more acidic pH.

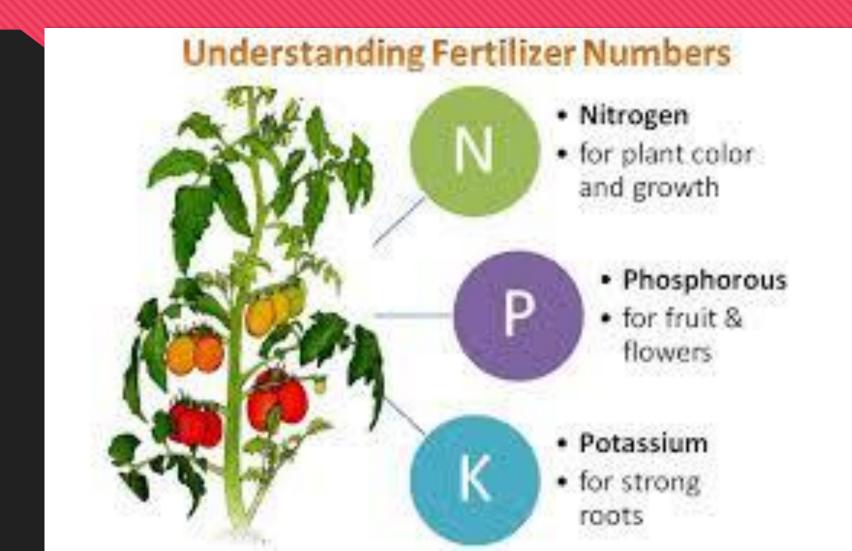
- OA pH test can be performed to determine the pH of the water, and amendments (pH up or pH down) can be added to modify the pH.
- Olimestone is commonly added to water in order to raise the pH.
- OSulfur or sulfur compounds can be added to the water to lower the pH.
- OIn our class we just use pH down.



- OWhen choosing Nutrients to add in your system, look for the fertilizer analysis on the container.
- OThe Nutrient analysis states the percentage of primary nutrients (nitrogen, phosphate, and potash) present.
- OThe analysis is written as three numbers (i.e., 15-10-26), which represent the percent of nitrogen, phosphorus, and potash present in the nutrient solution.
- OThey are always listed in that order.







- Olf a fertilizer contains all three primary nutrients, it is called a complete fertilizer.
- Olf a fertilizer is lacking any of the three primary nutrients, it is an incomplete fertilizer.
 - OThe fertilizer analysis does not equal 100%.
 - OThe rest of the fertilizer composition consists of filler materials (necessary for the fertilizer to be applied) and possibly some micronutrients.

Plants have different nutrient requirements based on the stage of growth.

- OThe first stage is when the plant is a young seedling or cutting.
- OIn the second stage, vegetative or leafy growth is encouraged.



- OThis is followed by the flower bud initiation stage, the flower bud development stage, and the flowering stages.
- OEach stage calls for different rates of fertilizers.
- OIn the vegetative stage, plants use more nitrogen, whereas the flowering stage requires less nitrogen and more phosphorus.



- OFertilizers are typically applied as water-soluble fertilizer or as slow-release fertilizers.
 - 1. Water-soluble fertilizers dissolve completely in water and stay in solution.
 - OThe concentrations of the water-soluble fertilizers are also easily adjusted.
 - OThe fertilizer concentrate is then mixed with the water in exact proportions.

- OAll the plants of a particular crop receive the same levels of nutrients, which assists in producing a uniform crop.
- OIn addition, adjustments to the level of nutrients in solution can be made easily.



- ONutrients in solution are measured in parts per million or PPM.
- OFertilizer rates can be taken from tables in reference books or to instructions with their injector system, or growers can calculate the amount of fertilizer needed to reach a desired PPM.
- OIn either case, it is beneficial for a grower to understand how to calculate parts per million (PPM) of fertilizer nutrients.